




BY Phd Student

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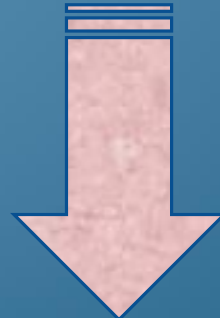
Prof. Dr. **FRIED MIKLOS**



**Combinatorial Preparation
and Characterization
Methods for High
Through-put Study of
Advanced Functional
Materials**

Aim of the Research

To understand and optimize the electrochromic behavior of mixed metal oxides deposited by reactive sputtering.



Research Work

We will prepare thin films of mixed Titanium, Tin Oxide and $\text{WO}_3\text{-MoO}_3$ mixed layers on glass by reactive DC magnetron sputtering. The deposited $\text{A}_x\text{B}_{1-x}\text{O}_n$ type films will be characterized by a variety of methods.

Research methods:

Preparation methods

Pulsed mode reactive DC magnetron , Biased RF sputtering systems (see Fig. 1) and Laser ablation deposition system

Characterization methods

Spectroscopic Ellipsometry, Rutherford Backscattering Spectrometry, Transmission Electron Microscopy, Scanning Electron Microscopy and Atomic Force Microscopy

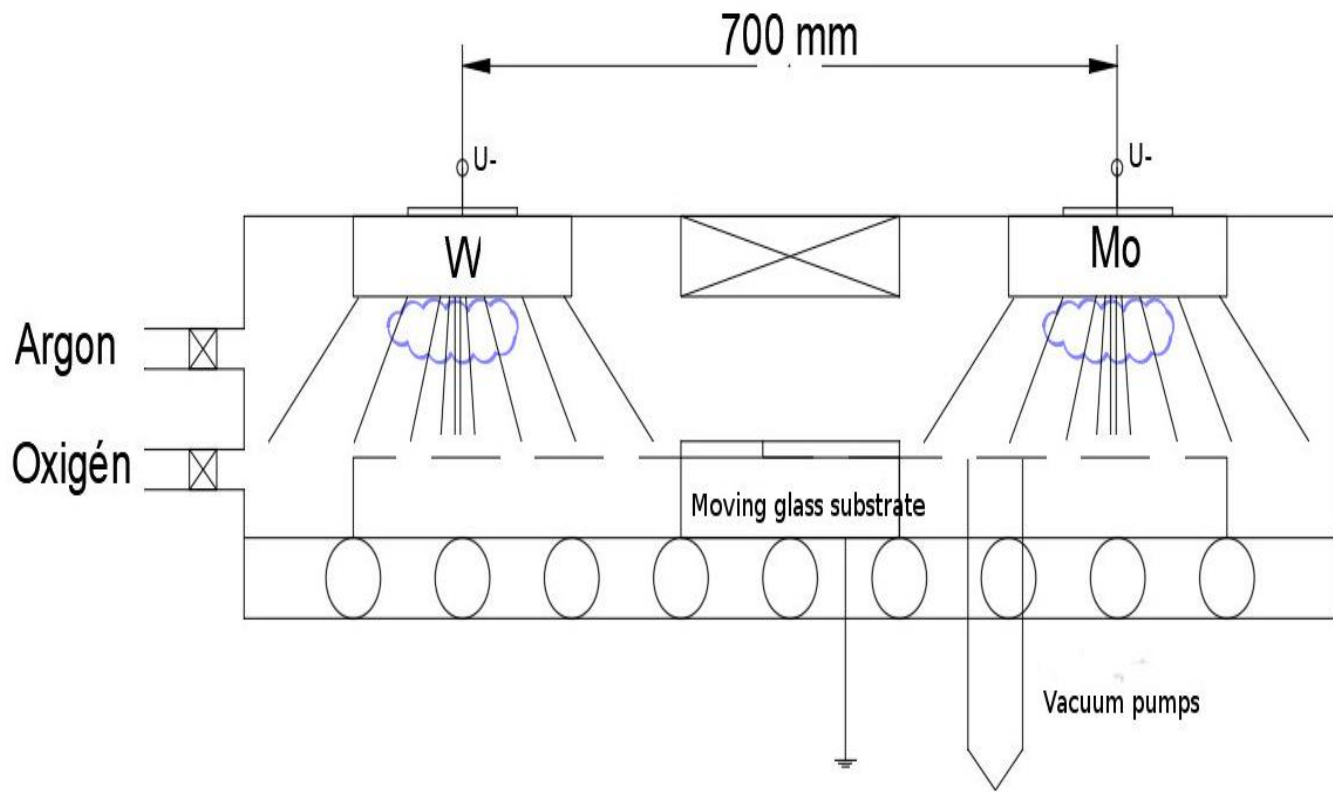


Figure (1) Schematic picture of the deposition arrangement and the photograph of 30x30 cm substrate with composition-gradient layers. Colored bands show thickness and composition gradient.

The research evaluations (The Result)

First half year we measured the WO_3/MoO_3 samples (like the one on Fig. 1) and fit the measurement by the CompleteEASE Wollam program shown in Figure (2).

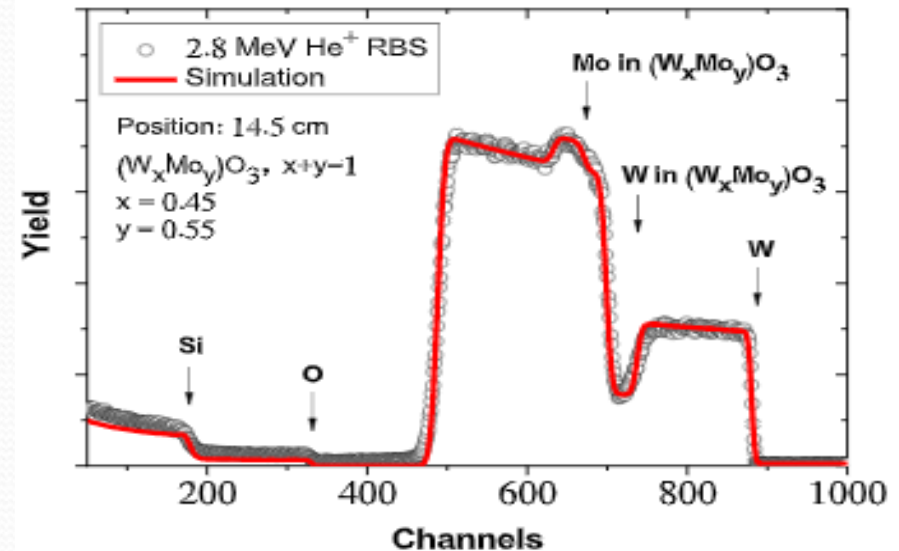
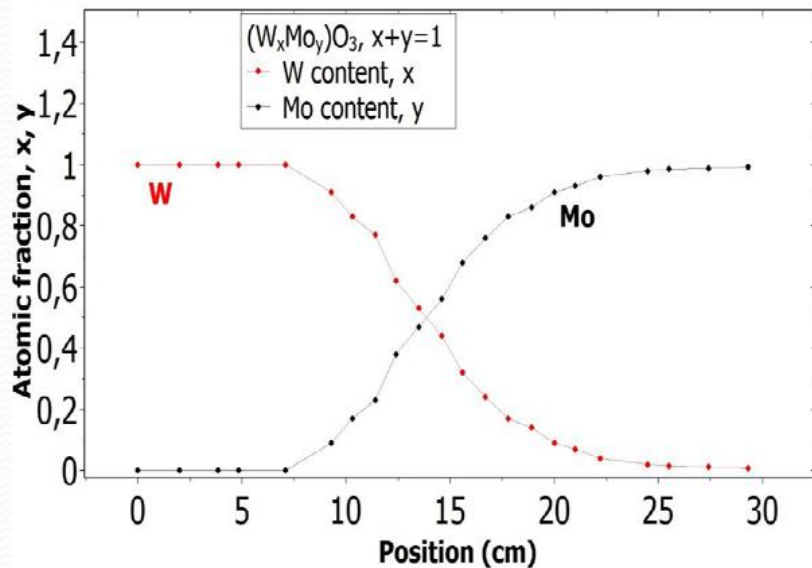


Fig. 2 a., Composition-map along a line by Rutherford Backscattering Spectrometry.

b., One Rutherford Backscattering Spectrometry example near the centre position.

Surf. roughness (air(50%) + $\text{WO}_3\text{-MoO}_3$ (50%))

$\text{WO}_3\text{-MoO}_3$ 200-1500 nm
2 T-L osc. or Effective Medium Approximation (EMA)

Interface (W(50%) + $\text{WO}_3\text{-MoO}_3$ (50%))

W on glass (substrate)

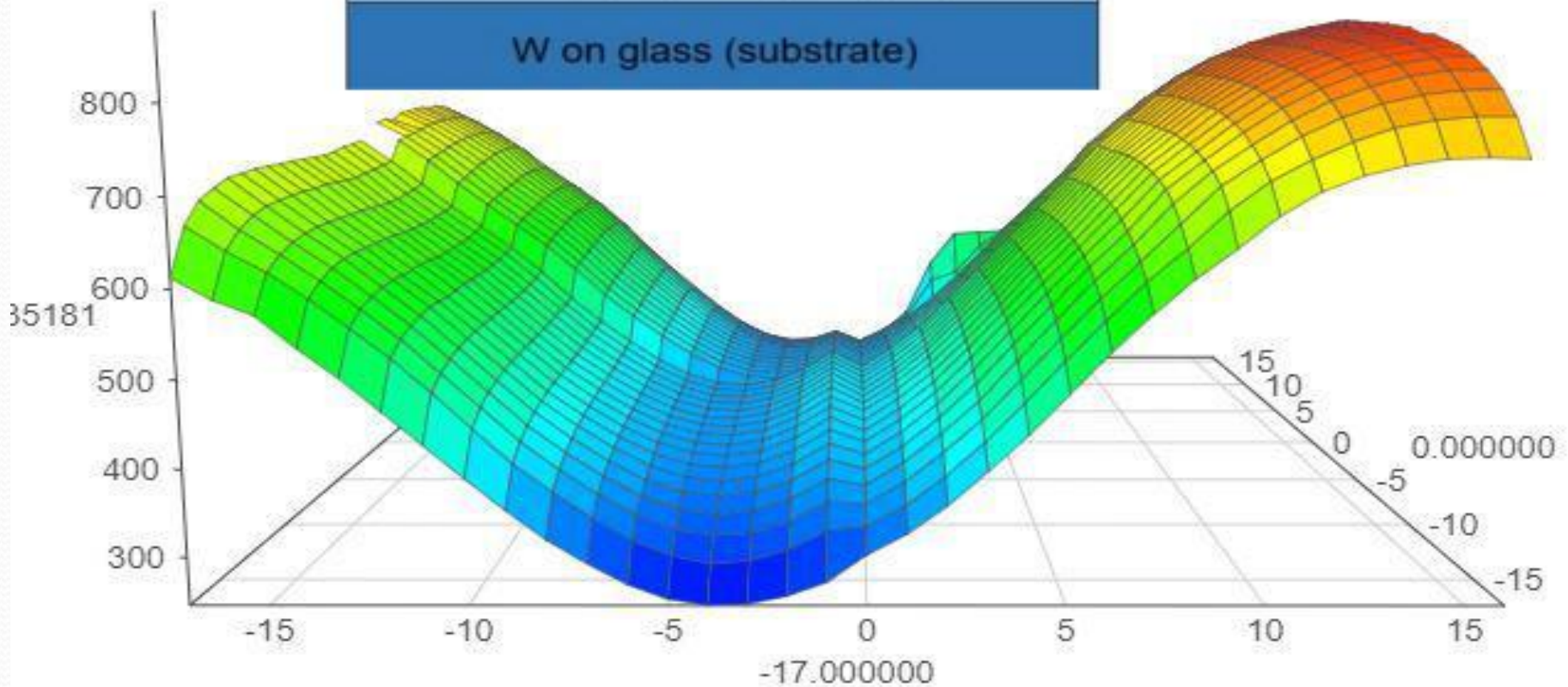


Fig. 2 c., Ellipsometric thickness mapping of the WO_3/MoO_3 combinatorial layer.

WO₃ by Cauchy Spectroscopic Data At X=-4.5, Y=0

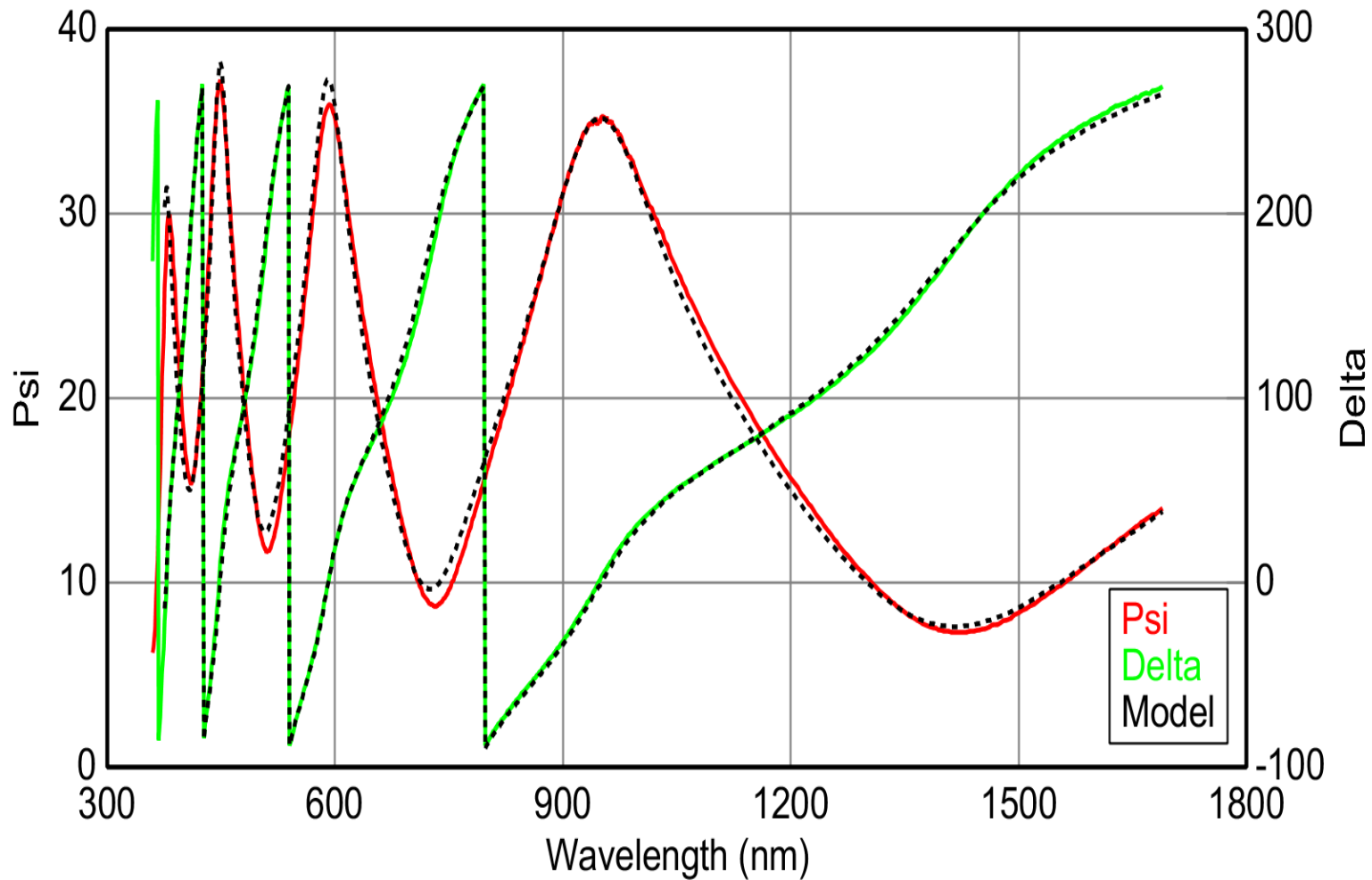


Figure (3) presented Ellipsometry mapping of the 100% WO₃ (left side of the sample) by Cauchy-Dispersion Formula Spectroscopic from (375-1700) nm.

- Layer # 2 = Cauchy Thickness # 2 = 386.61 nm (fit)
A = 2.041 (fit) B = 0.01603 (fit) C = 0.00450 (fit)
k Amplitude = 0.02065 (fit) Exponent = 3.146 (fit)
Band Edge = 400.0 nm

Layer # 1 = INTR_JAW Thickness # 1 = 1.00 nm

Substrate = SI_JAW

Angle Offset = -0.068

MSE = 25.2

A = 2.041 ± 0.002

B = 0.0160 ± 0.0007

C = 0.00450 ± 8.9924E-05

k Amplitude = 0.02065 ± 0.0006

Exponent = 3.15 ± 0.2

Total Thickness = 387.6 ± 0.5 nm.

wo3 tauc-lorenz Spectroscopic Data At X=-4.5, Y=0

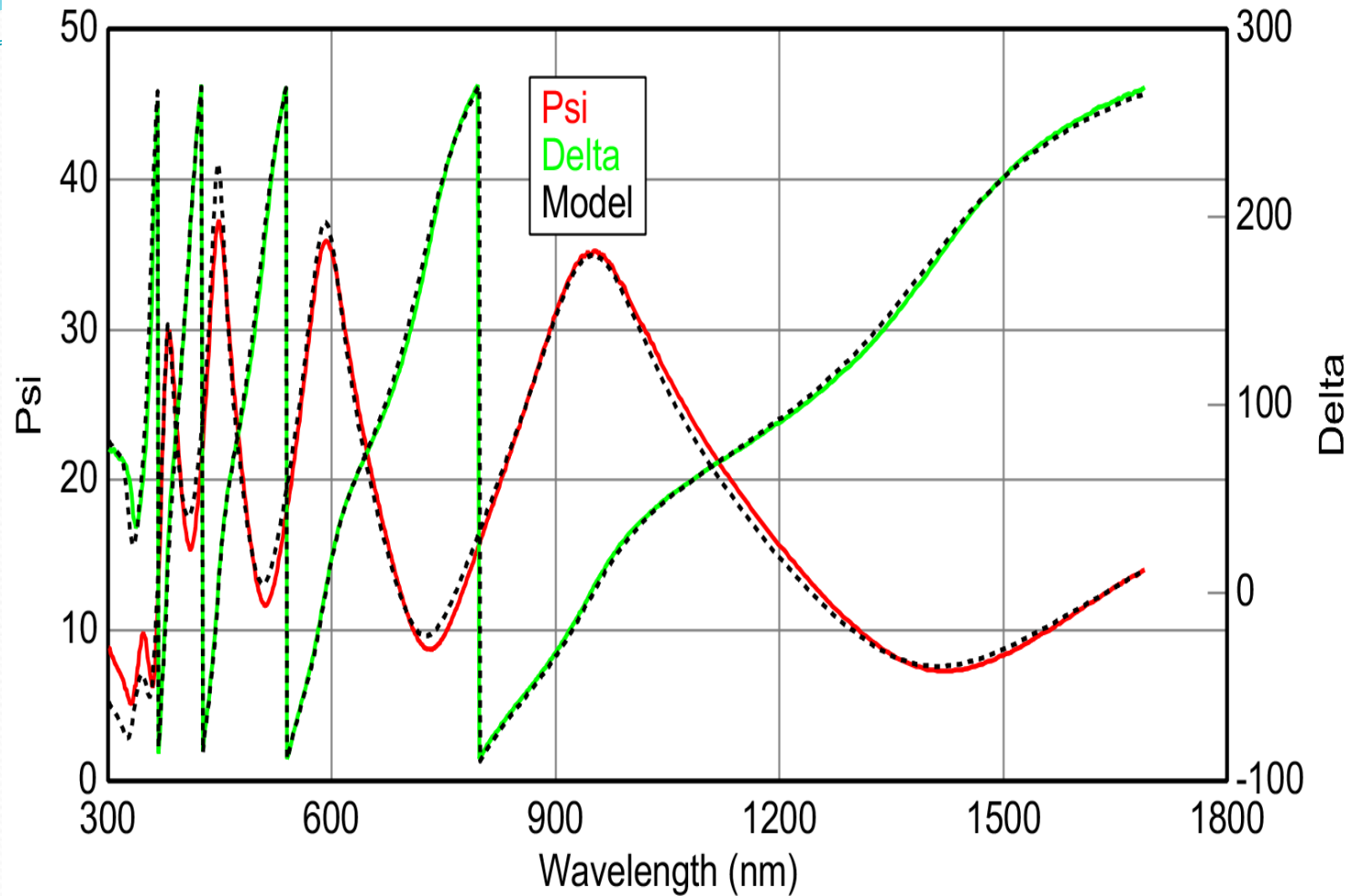


Figure (4) presented Ellipsometry mapping of the 100% WO₃ (left side of the sample) by Tauc-Lorenz Function Spectroscopic from (300-1700) nm.

MoO₃ Cauchy Spectroscopic Data At X=3.5, Y=0

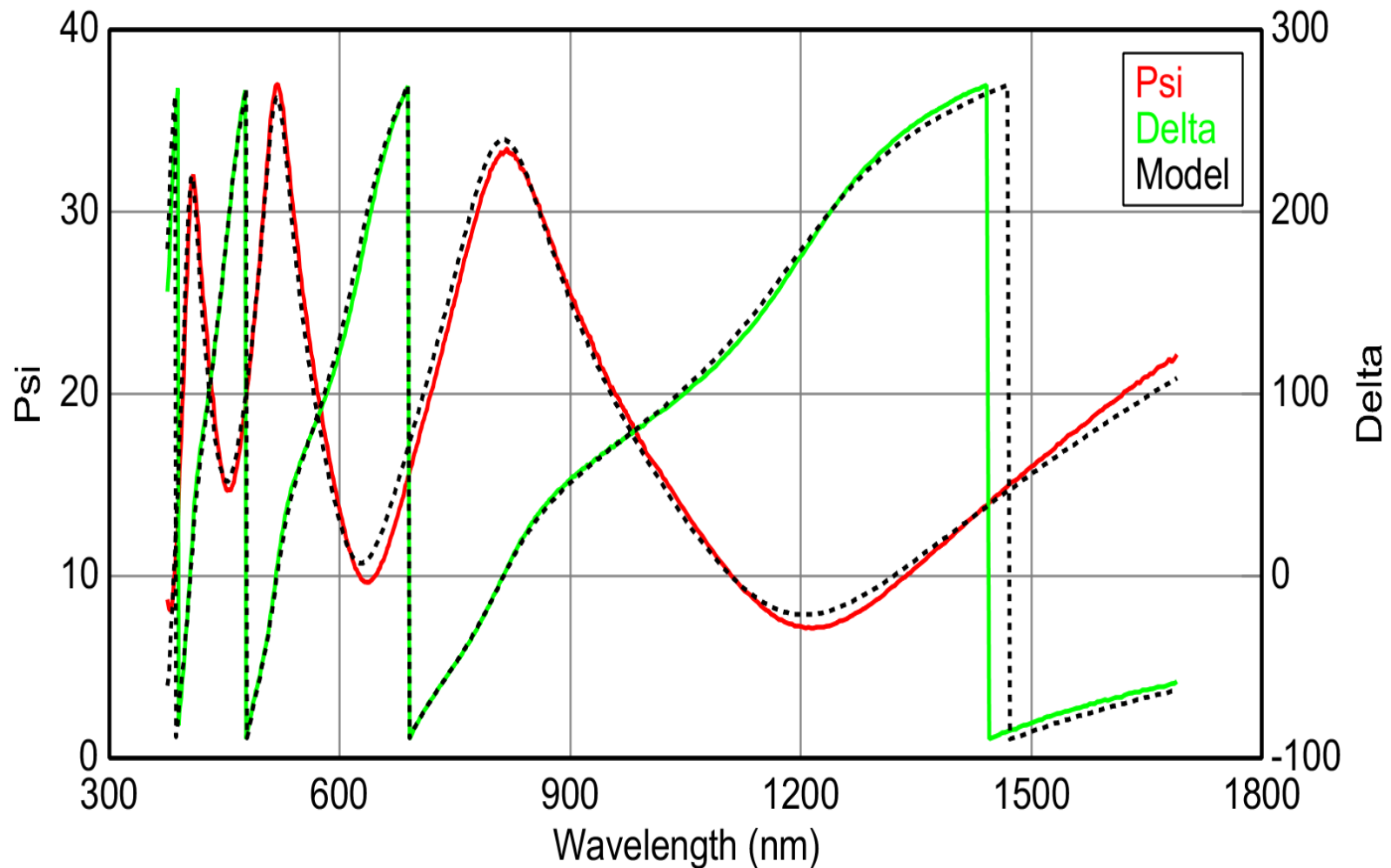


Figure (5) presented Ellipsometry mapping of the 100% MoO₃ (right side of the sample) by Cauchy-Dispersion Formula Spectroscopic from (375-1700) nm.

MoO₃ tAUC LORENZ Spectroscopic Data At X=3.5, Y=0

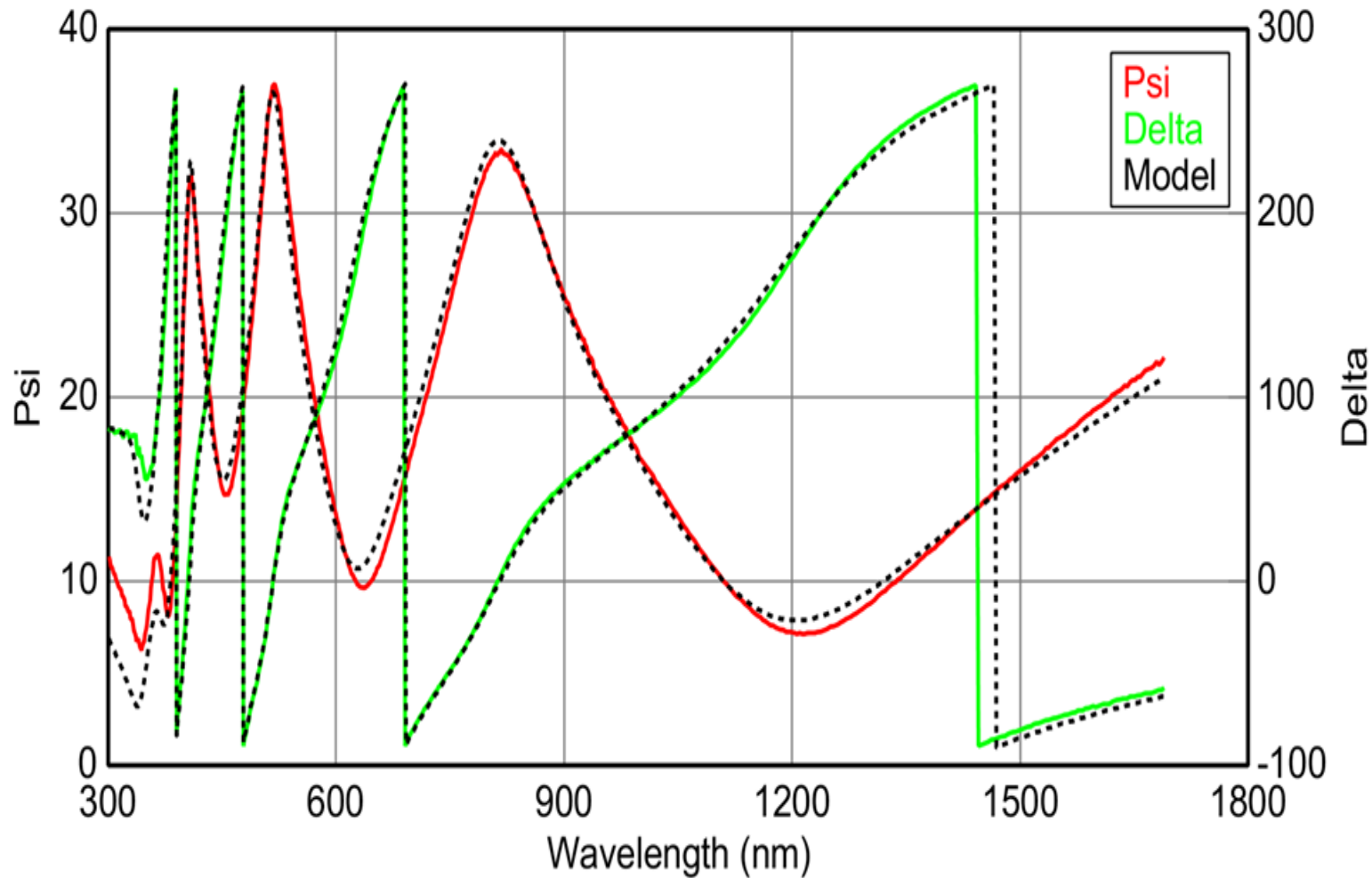


Figure (6) presented Ellipsometry mapping of the 100% MoO₃ (right side of the sample) by Tauc-Lorenz Function Spectroscopic from (300-1700) nm.

WO₃-MoO₃ EMA TAUC-LORENZ Spectroscopic Data At X=-1, Y=0

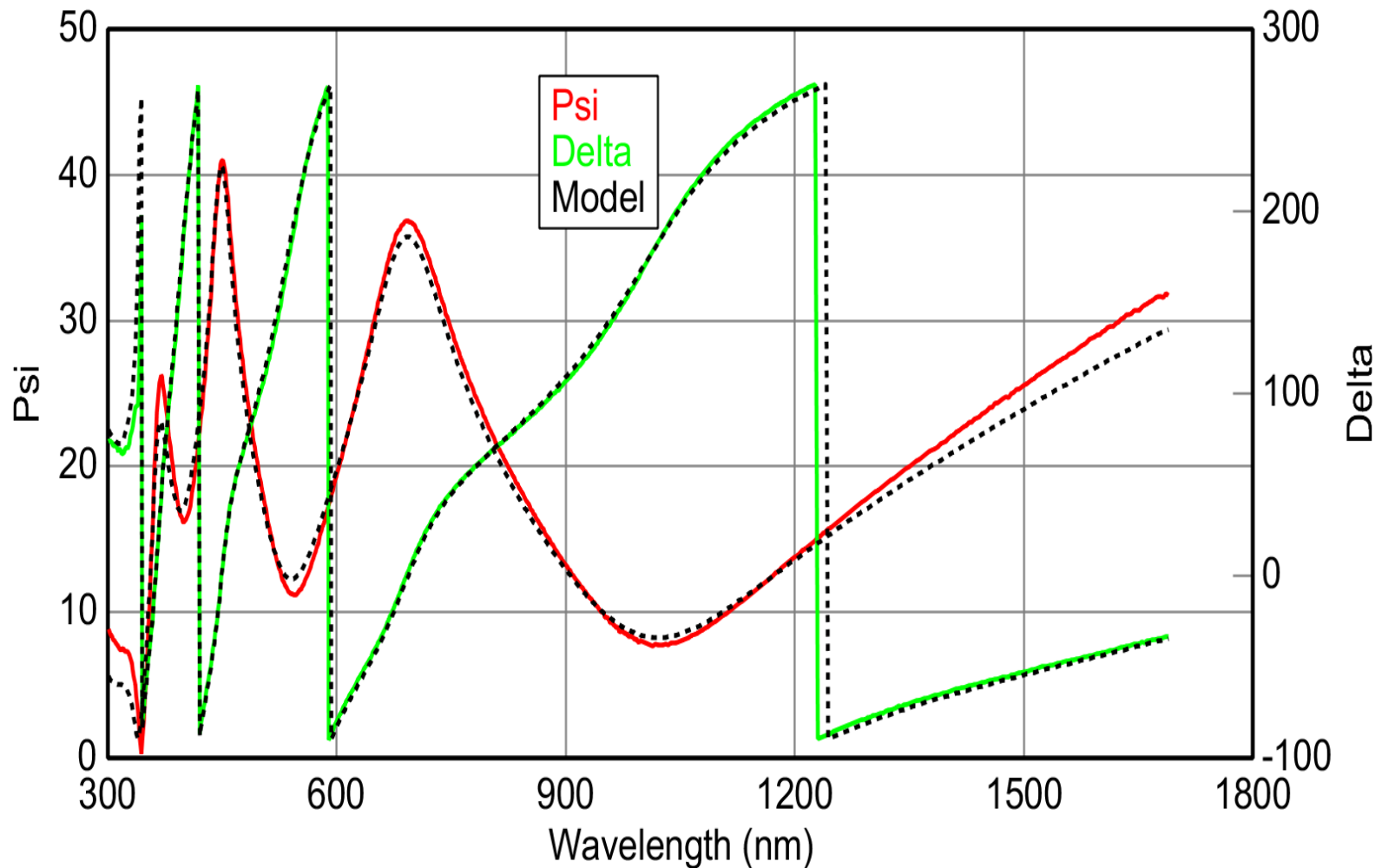


Figure (7) presented Ellipsometry mapping of the Sample: WO₃-MoO₃-on-10cm-long-siliconin by EMA Tauc-Lorenz Function Spectroscopic from (300-1700) nm.

EMA % (MoO₃) Thickness [nm] MSE vs Position [cm]

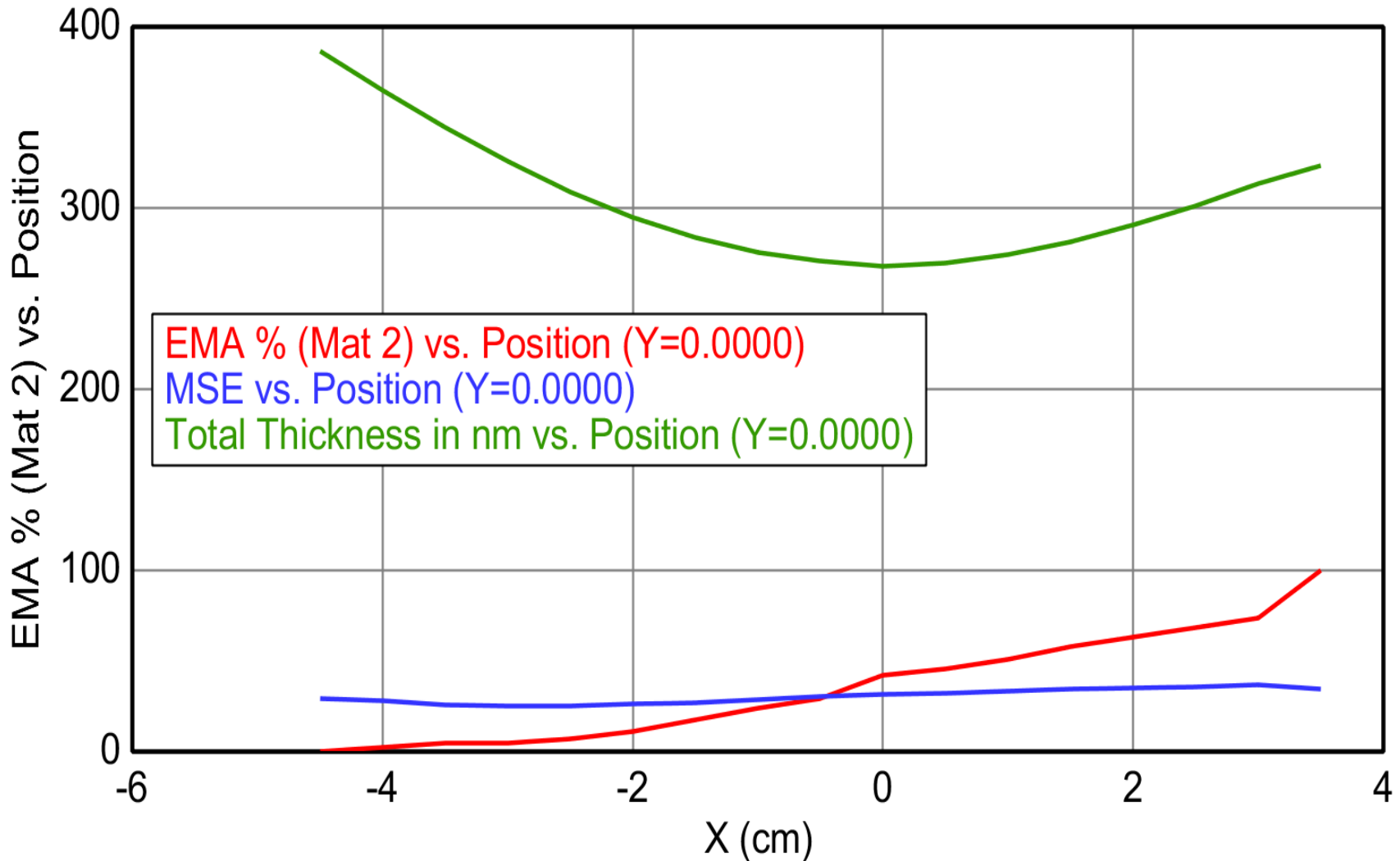


Figure (8) presented Ellipsometry mapping of the EMA % (MoO₃) Thickness (nm), MSE vs Position(cm).

Layer Commands: **Add** **Delete** **Save**

Include Surface Roughness = **OFF**

- Layer # 2 = **EMA** Thickness # 2 = **9906.98 nm** (fit)

of Constituents = **2**

+ Material 1 = **wo3 tauc-lorenz**

+ Material 2 = **MoO3 TAUC LORENZ**

EMA % (Mat 2) = **100.0** (fit)

depolarization = **0.333** Analysis Mode = **Bruggeman**

Layer # 1 = **INTR_JAW** Thickness # 1 = **1.00 nm**

Substrate = **SI_JAW**

Angle Offset = **-0.068**

+ **MODEL Options**

+ **FIT Options**

+ **OTHER Options**

Configure Options

Turn Off All Fit Parameters

• MSE = 598.049

• EMA % (Mat 2) = 100.0 ± 5.40

• Total Thickness = 9907. ± 10.870 nm.

A graphic featuring a wavy-edged banner with a rainbow gradient background. The banner is set against a solid blue background. The text "Plans for the future:" is written in a bold, black, italicized serif font, centered within the banner.

***Plans for the
future:***

The physical, chemical and structural properties of the cutting-edge materials are strongly dependent on their composition.

The common procedure to reveal the properties of concentration dependent phases is the preparation of numerous two (or more)-component samples, one for each $C_A/C_{B=1-A}$ composition, and the investigations of the individuals.

This is a low efficiency procedure that costs enormous time of man and machine.

Contrarily, using the combinatorial material synthesis approach, materials libraries can be produced in one experiment that contain up to several hundred or thousands of samples on a single substrate.

In order to identify optimized material structures in an efficient way, adequate automated micro-spot material characterization tools have to be applied.

Planned steps forward:-

- 1. Electrochromic measurements on stoichiometric WO_3/MoO_3 samples (Fig. 9).**
- 2. Further investigation of stoichiometric and sub-stoichiometric oxides for sensoric purposes (Fig. 10 and 11)).**
- 3. Ti and Tin Oxide (Sn oxides) combinatorial deposition (Fig. 12) and their measurements.**

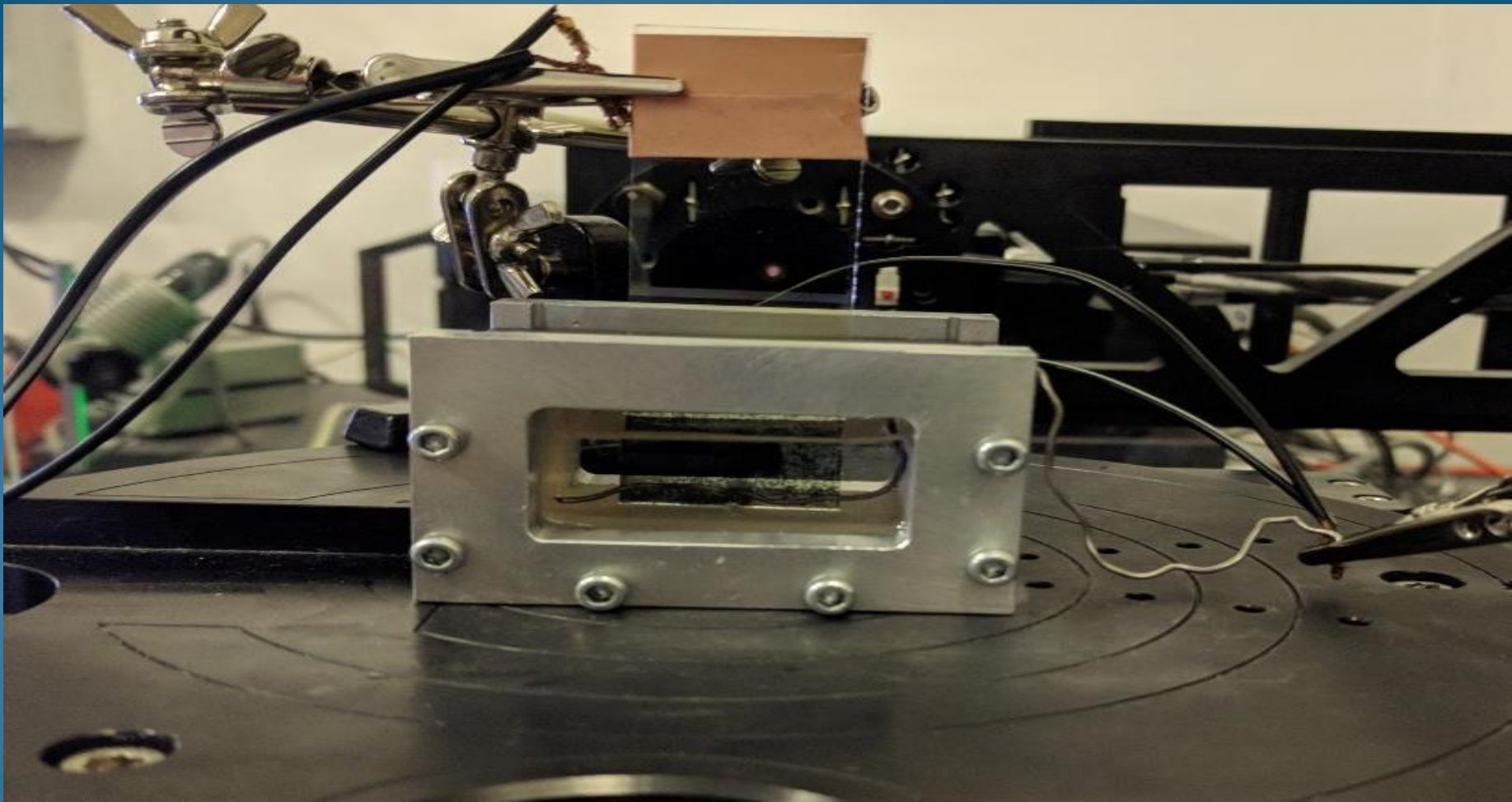


Figure (9) Liquid cell for electrochromic measurements in transmission mode



Fig. 10 Photographs (from different view-angle) of WO_3/MoO_3 (lower) or $\text{WO}_{3-x}/\text{MoO}_{3-x}$ (upper) combinatorial sets on heat-able sensor chips. Left hand side is W-rich, right one is Mo-rich. No. 6 (middle one) is expected to be 50-50% in both cases. The upper rows show sub oxides (semi-transparent layers, No. 2 was broken during tweezer handling) The bottom rows show oxides (transparent layers).

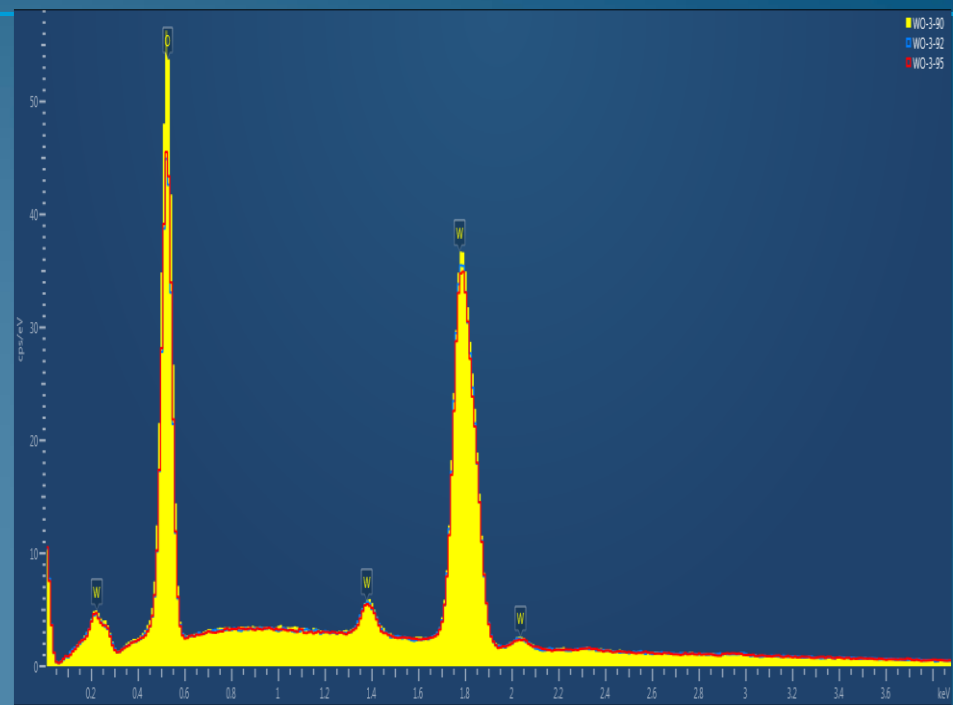
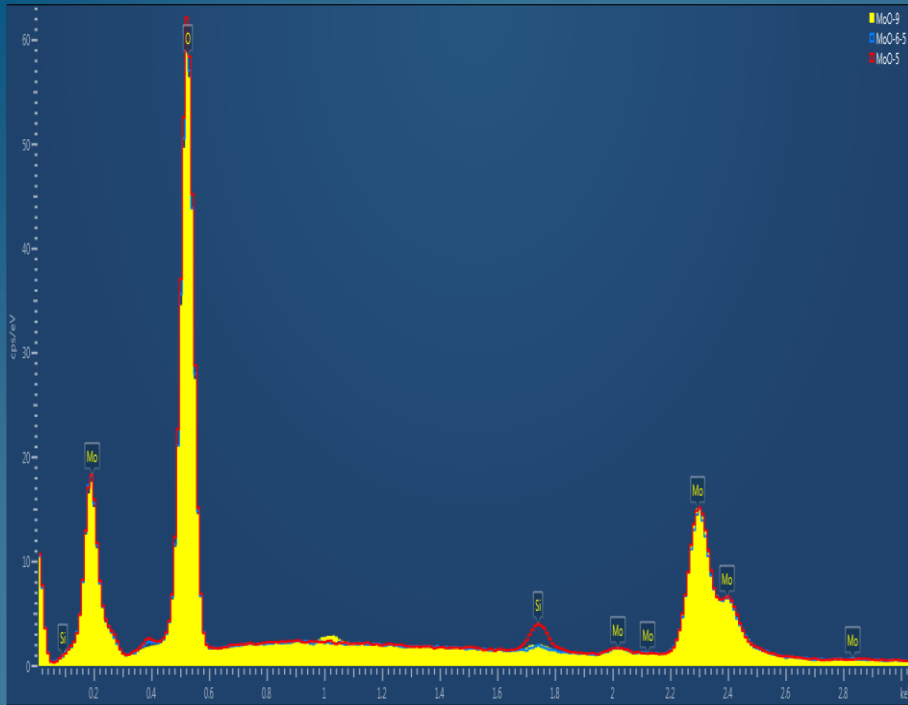


Fig. 11 EDS spectra of stoichiometric oxides and suboxides of W and Mo Semi-quantitative analysis shows 10% oxygen vacancy.



Fig. 12 *Ti and Sn oxides combinatorial deposition. Silicon probes to measure the thickness and composition map.*

*Subjects fulfilled
in the semester:*





Thanks for your attention