



Óbuda University, Doctoral School on Material Science  
and Technology

# "Synthesis and Application of Organic-Inorganic Nanocomposites in Artificial Photosynthesis"

8<sup>th</sup> semester

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# Outline

## ▶ Introduction:

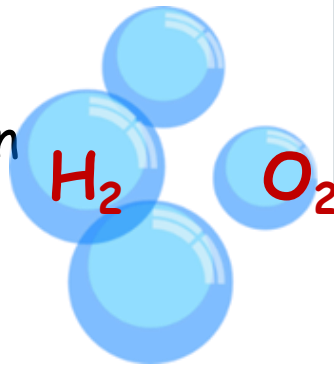
- ❑ Solar to Fuels (STF), H<sub>2</sub> Fuel.
- ❑ Photosynthesis: Natural & Artificial Photosynthesis (AP)
- ❑ Water splitting: Oxygen Evolution Reaction (OER) & Hydrogen Evolution Reaction (HER)

## Objectives

- ❑ Find a Robust, Efficient, Cheap catalysts for **Water Oxidation (WO)**

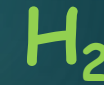
## ▶ Publications

## ▶ Thesis points



# Solar to Chemical Energy

(Fuel)



## Global Energy

- Coal,
- Oil products, Natural gas

## Provided

- Carbon-based energy
- CO<sub>2</sub> emission

## Environment

- Air pollution
- Green house



Hydrogen by Photoelectrochemical cells (PEC)



Solar cells only generating during day.



Renewable Resources Available, Ex: Wind, Hydro. Power, **Solar**...etc

Fossil Fuels being depleted, effect climate, CO<sub>2</sub>

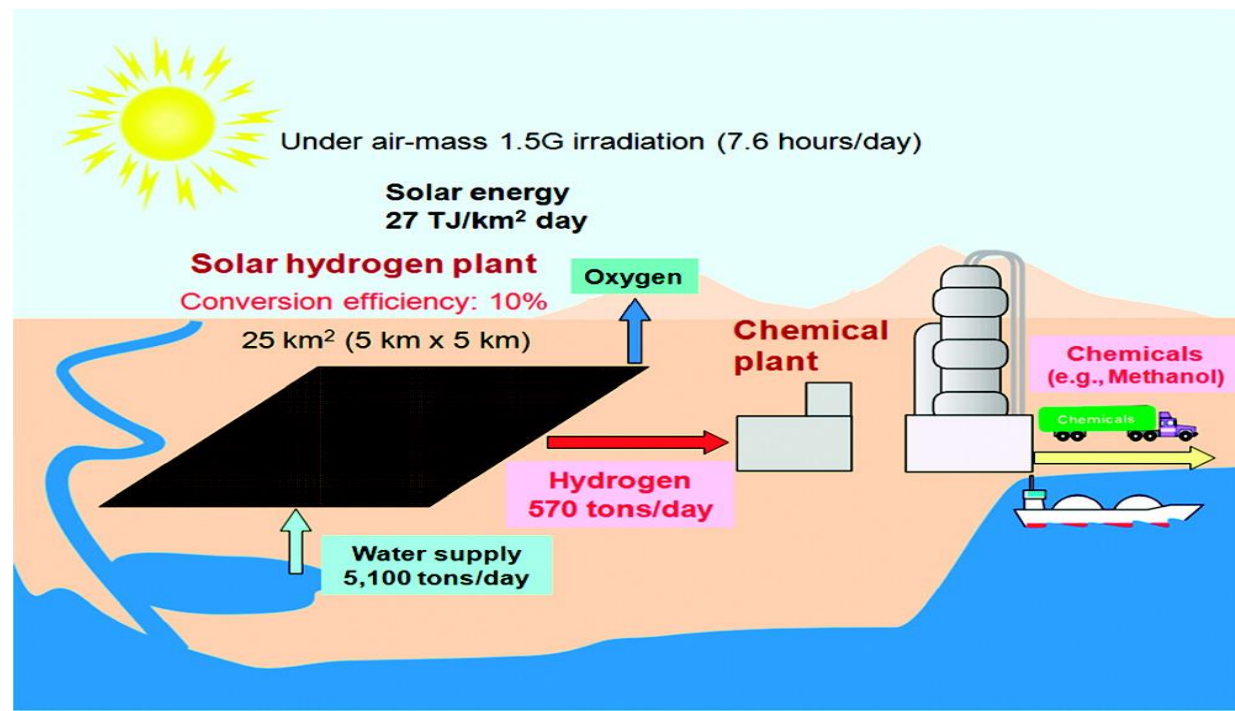
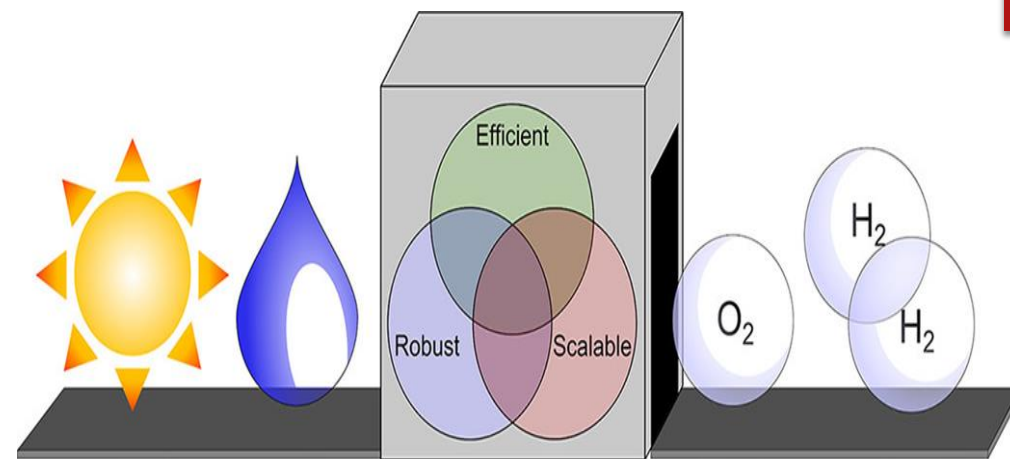


# Solar to Chemical Energy-Fuel (STF)

How can achieve STF by water splitting?

- In an artificial photosynthetic system, essential to devise an efficient process,
- need to be made from cheap and abundant materials.

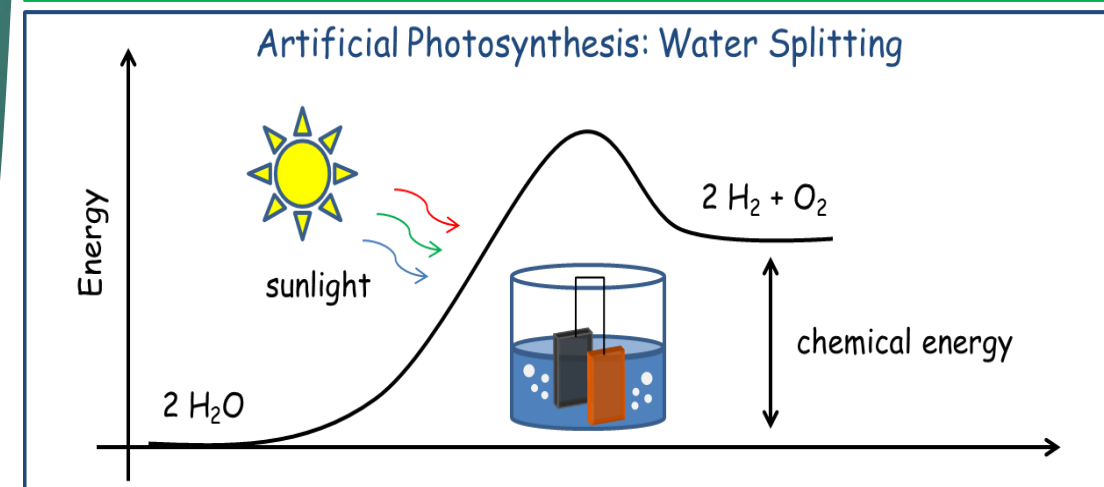
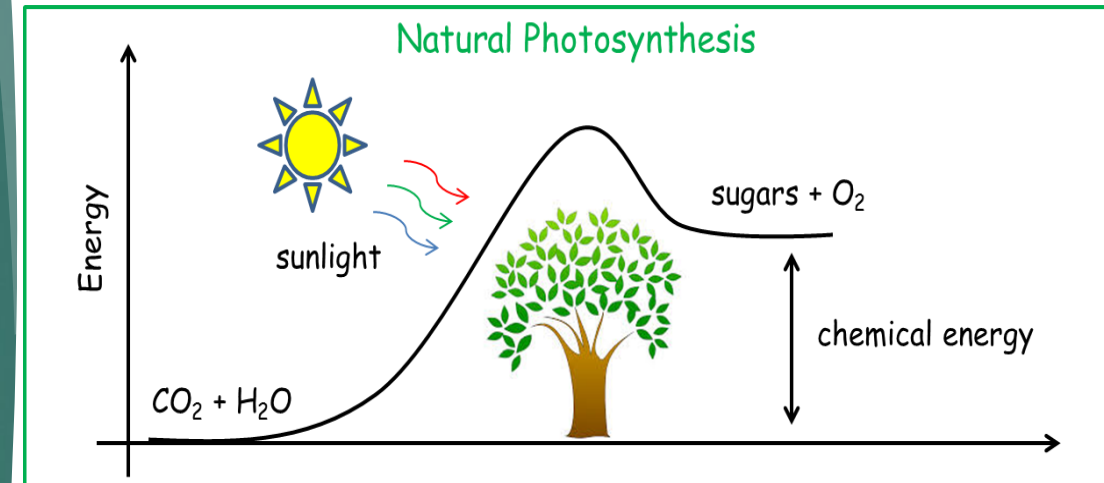
Scheme shows the possible large-scale  $H_2$  production via solar water splitting.



# Photosynthesis: Natural vs Artificial

☐ Photosynthesis has produced most of the energy that sustains life on our planet

- ☐ The artificial photosynthesis aims to mimic the natural process
- ☐ Conversion of sunlight into  $H_2$  and  $O_2$  by solar-driven water splitting



# Artificial photosynthesis

In this machinery

five major processes:  
photon absorption,  
charge separation,  
electron transfer,  
water oxidation, and  
proton reduction, must be combined  
to achieve high efficiency.

## Water oxidation (WO)

the bottleneck in the field of  
electrochemical

# Electrochemical Water splitting

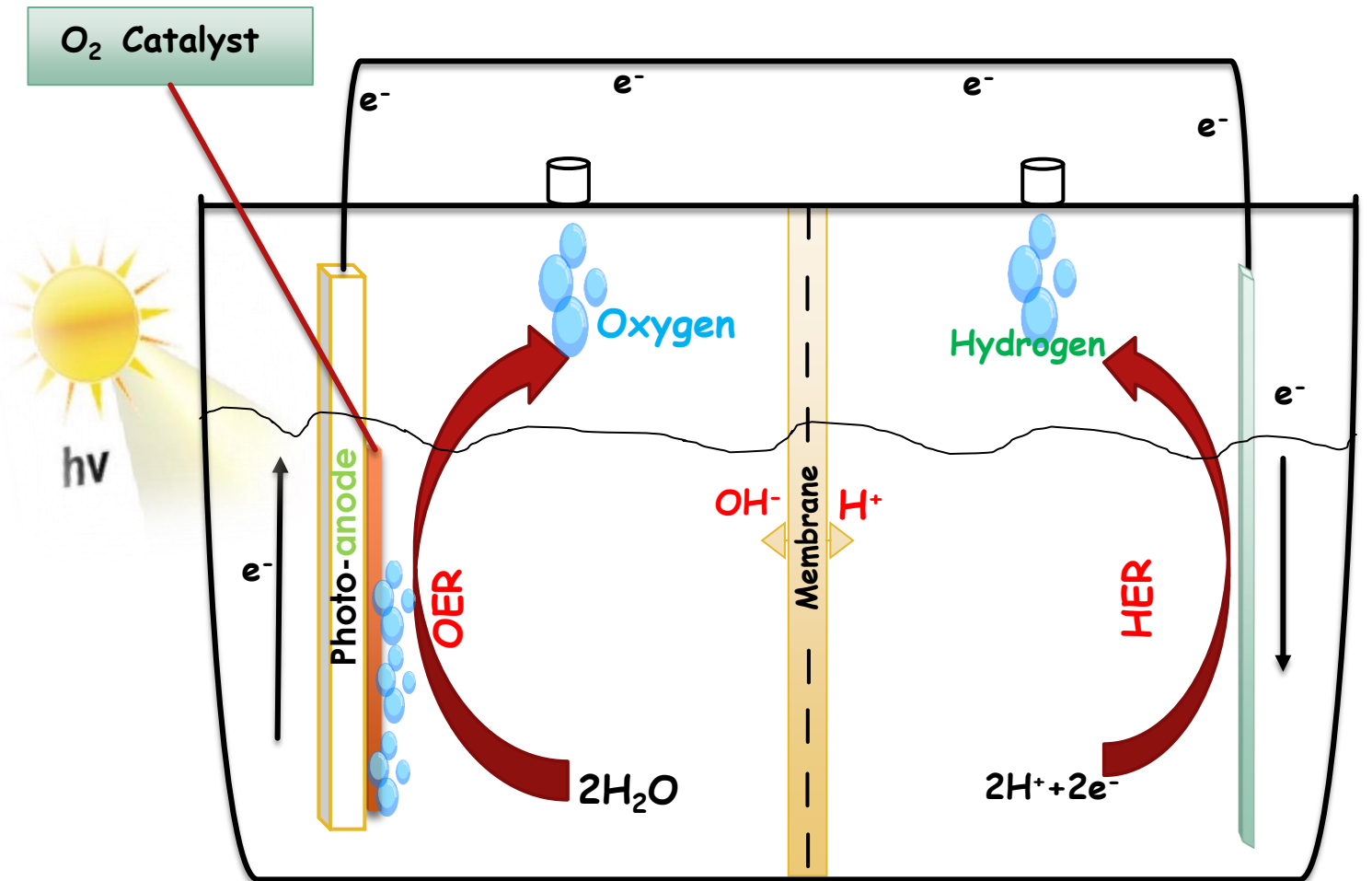


Figure: An illustration of a type of PEC device

# objectives

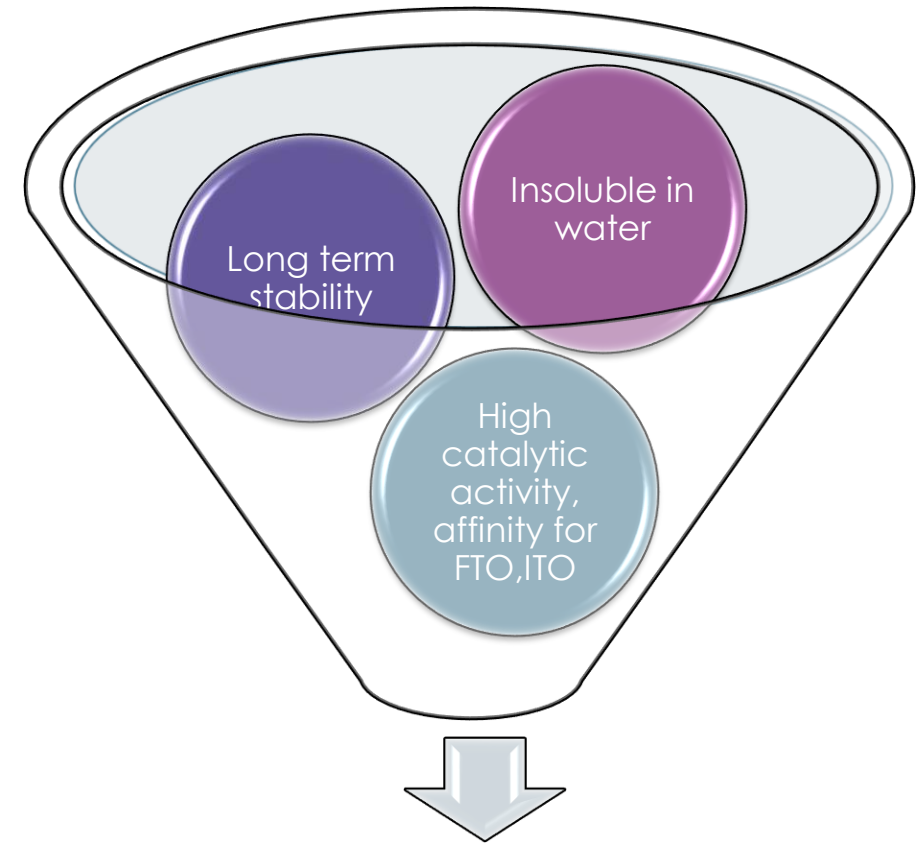
**Catalysts** are a very important component in efforts to design and develop efficient water splitting technologies.

Our efforts, like in many other research groups, is directed at the development of Molecular catalysts,

An **efficient** and **robust** catalyst for WO based on:

□ abundant and cheap materials

the key to converting solar energy into fuels through **artificial photosynthesis**.



Applied in electrocatalytic  
**Water Oxidation (WO)**



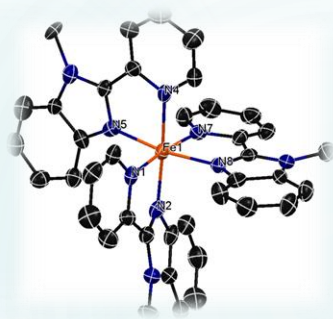
1<sup>st</sup> publication

Journal of Catalysis open access IF 7.888

# Utilization of hydrophobic ligands for water-insoluble Fe(II) water oxidation catalysts - Immobilization and characterization

We Compared Two Fe<sup>II</sup> complexes by electrochemical methods (homogeneous and heterogeneous conditions)

Fe<sup>(II)</sup>(PBI)OTf



PBI

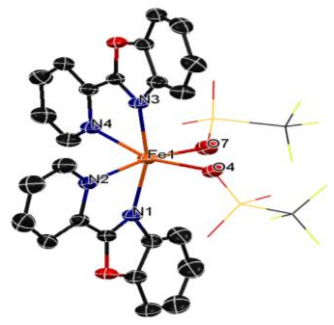


PBI

drop-casting on ITO

PBO

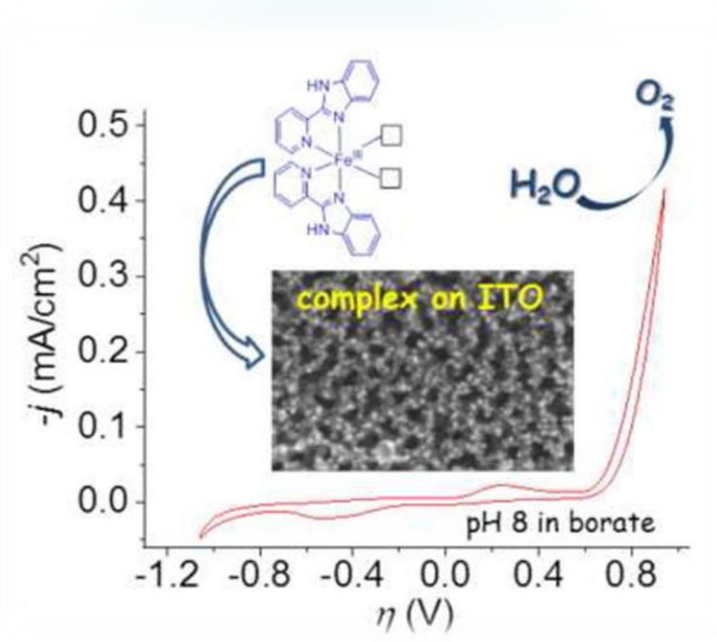
Fe<sup>(II)</sup>(PBO)OTf



PBO



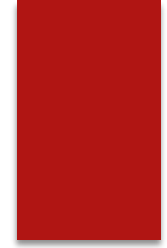
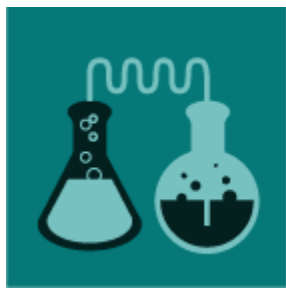
Surface deposition



Ligands:  
2-(2'-pyridyl)benzimidazole (PBI)  
2-(2'-pyridyl)benzoxazole (PBO)

OTf = trifluoromethyl sulfonate anion

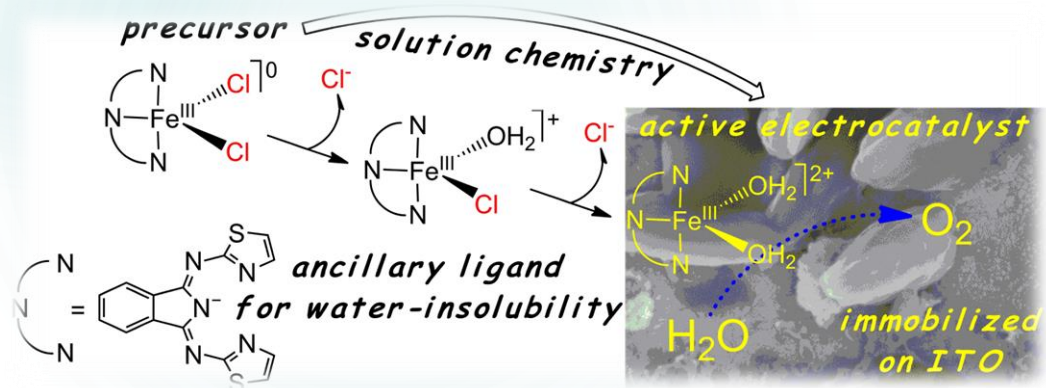




**“An Iron(III) Complex with Pincer Ligand—Catalytic Water Oxidation through Controllable Ligand Exchange”**

- we successfully synthesized and investigated the electrochemical properties of the five-coordinate complex  $[\text{Fe}^{\text{III}}\text{Cl}_2(\text{tia-ind})]$ , as a potential pre-catalyst of water oxidation, in homogeneous water/acetone mixture to reveal the signatures of  $\text{Cl}^-$  to  $\text{H}_2\text{O}$  ligand exchange

**The five-coordinate complex  $[\text{Fe}^{\text{III}}\text{Cl}_2(\text{tia-ind})]$**



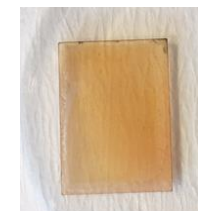
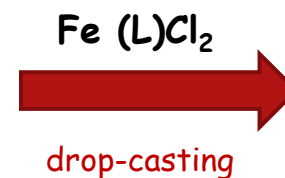
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$L$ , tia-ind = 1,3-bis(2'-thiazolylimino)isoindolate(-)

- Immobilization of the complex from methanol on indium-tin-oxide (ITO) electrode by drop-casting resulted in water oxidation catalysis in borate buffer



ITO





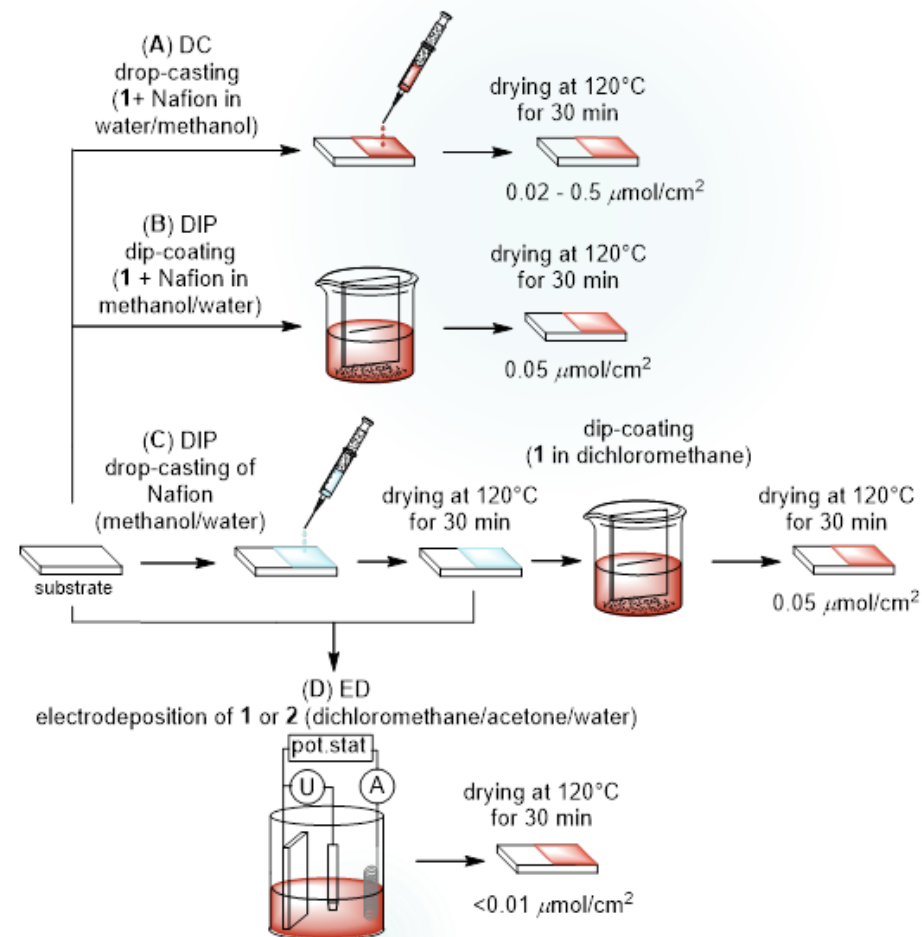
## “Redox-Driven Electrodeposition of Fe-Complexes on Oxide Surfaces for Efficient OER Catalysis”

□ We study the electrochemical properties of two Fe<sup>(III)</sup> complexes with NN'N pincer ligands in homogeneous water/Acetone/DCM mixture to reveal the signatures of Cl<sup>-</sup> to H<sub>2</sub>O ligand exchange

□ strategies to enhance the performance of molecular catalysts: **dip-coating**, **drop-casting**, and **electrodeposition**, important for design for the immobilized catalyst exhibits much higher activity.

**Electrodeposition:** Simplicity, low cost, mild operating conditions, scalability

□ two Fe<sup>(III)</sup> complexes that prepared with NN'N pincer ligands yielding by redox-driven electrodeposition (ED) of stable and active ad-layers for the electrocatalysis of the oxygen evolving reaction (OER).

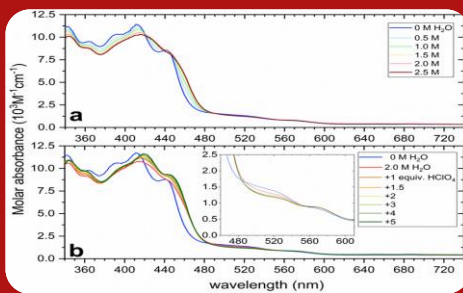


Different characteristics techniques have been used to The investigation of the complex/ITO or FTO assembly before and after catalysis that suggested that a molecular form of catalyst is responsible for water oxidation.

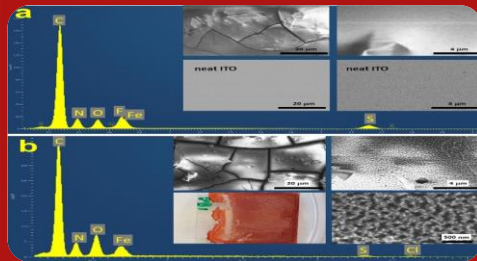
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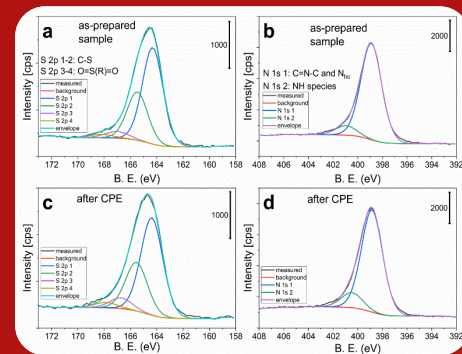
### Uv-vis spectrophotometer



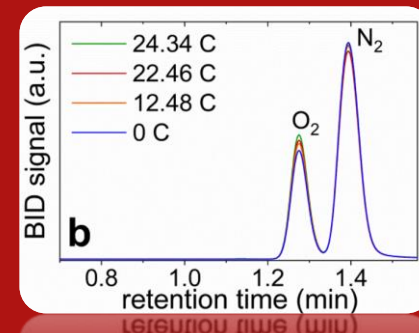
### SEM & EDX



### Xray photoelectron spectroscopy (XPS)



### Gas Chromatography (GC) to detect Oxygen evolution



# thesis points based on publications

- ❑ The water-insolubility of the complexes seems to be a viable strategy for the design of new molecular catalyst/(photo)anode hybrids (Hydrophobic ligands).
- ❑ Ligand exchange reaction allows for the active form of the molecular catalyst methanol that promotes ligand exchange the selection of the solvent for *drop-casting* is of key importance to gain an active heterogenized WOC.
- ❑ The non-coordinated heteroatoms in a heterocyclic ligand can induce fundamental changes in the redox behavior both in the homogeneous water/acetonitrile phase and when the complexes are deposited to the electrode and tested as water oxidation catalysts.
- ❑ No anchoring additive is needed, hybrid systems that are efficient in water oxidation electrocatalysis can be fabricated by the scalable and simple drop-casting method.
- ❑ The surface modification by Electrodeposition (ED) compared to other methods to fabricate ad-layers, including dip-coating, drop-casting is considered material saving and provides more efficient OER catalysis.



Thank you!